

The 6th International Conference on Nanogenerators and Piezotronics



hosted by FSCN reseach centre Mid Sweden University Sundsvall



Enhancement on Degradation of Organic Dye through Piezophotocatalytic Activities by High Entropy Oxide- $(CaZrYCeCr)O_2 / Bi_4Ti_3O_{12}$ Nanocomposite

Shun Cheng Chang¹, [#]Jyh Ming Wu¹

Department of Materials Science and Engineering, National Tsing Hua University

Abstract

High entropy material demonstrates several effects from multi-component composition, which is potential in the catalytical field. During this

decade, combining the mechanical force-driven piezoelectric/ferroelectric catalysts with photocatalysts has attracted much attention due to the separation of photogenerated carriers, leading to synergistic catalytic activities. This study reports a novel composite of high entropy oxide $(CaZrYCeCr)O_2$ with unique ferroelectric properties mixing with bismuth titanate $(Bi_4Ti_3O_{12})$ microplates, improving the piezophototronic catalytic activities. The results show that high entropy oxide catalyst (CaZrYCeCr)O₂ mainly consists of the porous structure with multivalent transition metal substitution, contributing to severe lattice distortion. Consequently, the precipitation of disordered phase $Ca(Zr_{1-x}Cr_x)O_3(x=0.05\sim0.1)$ caused by lattice distortion was decorated on the porous framework. In addition, $Ca(Zr_{1-x}Cr_x)O_3$ with a noncentrosymmetric structure exhibits ferroelectric property and plays a vital role in separating the electron-hole pairs, which performs higher degradation efficiency around 588 % compared to pristine ZrYCeO₂ Additionally, the ferroelectric Bi₄Ti₃O₁₂ was considered to construct a Z-scheme catalytic activity and form a heterojunction high entropy oxide. Furthermore, we demonstrated that morphotropic phase boundaries between $Bi_4Ti_3O_{12}$ and $CaZr_{1-x}Cr_xO_3$ could dramatically enhance the ferroelectric property and promote the performance of the piezo-phototronic activities, revealing that CaZrYCeCrO₂/ $Bi_4Ti_3O_{12}$ could be an innovative catalyst in the future.

Experimental process

a. Synthesis of high entropy oxide





Result and discussion

Figure 2. (a) P-E hysteresis loop of HEO with different content of calcium. (b) The schematic diagram of the ZrO_6 and CrO_6 in the $CaZr_{1-x}Cr_{x}O_{3}$ crystal structure and its deviation from the octahedral center. (c) EPR spectra of HEO with different content of calcium. (d) The corresponding result of the three stages varied by calcium content derived from (a,c). (e) A

b. Preparation of bismuth titanate



c. Preparation of high entropy oxide /bismuth titanate nanocomposite



Figure 1. (a,b) XRD pattern of high entropy oxide(HEO) with different calcium content(ex: 9HEO means HEO with 9at% Ca, AI for calibration) and $38\text{HEO/Bi}_{4}\text{Ti}_{3}\text{O}_{12}$ nanocomposite (c) SEM images of the 38HEO and (d) bismuth titanate. (e) TEM EDS mapping of 38HEO. (f) Low-magnitude TEM images of 38HEO. (g) HRTEM image of 9HEO with fluorite structure showing its (111) plane. (h) HRTEM image of 38HEO showing the (i) (200) plane of $CaZr_{1-x}Cr_{x}O_{3}$ (j) (111) plane of $CaZrYCeCrO_{2-x}$ and their SAED pattern.







Figure 3. Dye degradation of methylene blue using HEO under (a) applying light irradiation only, (b) applying ultrasonic vibration only and (c) applying light and ultrasonic vibration simultaneously.

Conclusion

- 1. As increasing Ca content, ferroelectric perovskite structure will precipitate from fluorite structure.
- 2. HEO with 38at% Ca has the optimal hysteresis loop with the vacancy concentration, resulting in the highest efficiency in methylene